

FORM PTO-1390
(REV. 6-87)U.S. DEPARTMENT OF COMMERCE
PATENT AND TRADEMARK OFFICEATTORNEY'S DOCKET NUMBER
1205-01**TRANSMITTAL LETTER TO THE UNITED STATES
DESIGNATED/ELECTED OFFICE (DO/EO/US)**

09/868480

INTERNATIONAL APPLICATION NO. PCT/JP00/07269	INTERNATIONAL FILING DATE 19 October 2000 (19.10.00)	PRIORITY DATE CLAIMED 20 October 1999 (20.10.99)
---	---	---

TITLE OF INVENTION

BLAST FURNACE COKE HAVING HIGH REACTIVITY AND HIGH STRENGTH AND METHOD OF PRODUCING THE SAME

APPLICANT(S) FOR DO/EO/US

Koji Hanaoka et al.

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items under 35 U.S.C. 371:

1. This express request to immediately begin national examination procedures (35 U.S.C. 371(f)).
2. The U.S. National Fee (35 U.S.C. 371(c)(1)) and other fees as follows:

CLAIMS	(1) FOR	(2) NUMBER FILED	(3) NUMBER EXTRA	(4) RATE	(5) CALCULATIONS
	TOTAL CLAIMS	12 -2=	0	x \$18.00	\$
	INDEPENDENT CLAIMS	4 -3=	1	x \$80.00	80.00
	MULTIPLE DEPENDENT CLAIM(S) (if applicable)			+ \$270.00	270.00
	BASIC NATIONAL FEE (37 CFR 1.492(a)(1)-(4)):				
	<input type="checkbox"/> International preliminary examination fee paid to USPTO (37 CFR 1.482).....			\$690.00	
	<input type="checkbox"/> No international preliminary examination fee paid to USPTO (37 CFR 1.482) but international search fee paid to USPTO (37 CFR 1.445(a)(2)).....			\$710.00	
	<input type="checkbox"/> Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO.....			\$1,000.00	
	<input type="checkbox"/> International preliminary examination fee paid to USPTO (37 CFR 1.482) and all claims satisfied provisions of PCT Article 33(2) to (4).....			\$ 100.00	
	<input checked="" type="checkbox"/> International Search Report enclosed			\$860.00	860.00
	Surcharge of \$ _____ for furnishing the National fee or oath or declaration later than <input type="checkbox"/> 20 <input type="checkbox"/> 30 mos. from the earliest claimed priority date (37 CFR 1.482(e)).			\$130.00	
				TOTAL OF ABOVE CALCULATIONS	\$1,210.00
	Reduction by ½ for filing by small entity, if applicable. Affidavits must be filed also. (Note 37 CFR 1.9, 1.27, 1.28.)				
				SUBTOTAL	\$1,210.00
	Processing fee of \$ _____ for furnishing the English Translation later than <input type="checkbox"/> 20 <input type="checkbox"/> 30 mos. from the earliest claimed priority date (37 CFR 1.482(f)).			\$130.00	
				TOTAL NATIONAL FEE	\$1,210.00
	Fee for recording the enclosed assignment (37 CFR 1.21(h)).			\$40.00	+
				TOTAL FEES ENCLOSED	\$1,210.00

- a. A check in the amount of \$1,210.00 to cover the above fees is enclosed.
- b. Please charge my Deposit Account No. 13-3405 in the amount of \$ _____ to cover the above fees.
A duplicate copy of this sheet is enclosed.
- c. The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 13-3405. A duplicate copy of this sheet is enclosed.

09/868480

3. A copy of the International Application as filed (35 U.S.C. 371(c)(2))
 - a. is transmitted herewith (required only if not transmitted by the International Bureau).
 - b. is not required, as the application was filed in the United States Receiving Office (RO/US).
 - c. has been transmitted by the International Bureau.
4. A translation of the International Application into English (35 U.S.C. 371(c)(2)).
5. Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371(c)(3))
 - a. are transmitted herewith (required only if not transmitted by the International Bureau).
 - b. have been transmitted by the International Bureau.
6. A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
7. An oath or declaration of the inventor (35 U.S.C. 371(c)(4)).
8. A translation of the Annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)).

Other document(s) or information included:

9. An Information Disclosure Statement under 37 C.F.R. 1.97 and 1.98.
10. An Assignment document for recording and a Recordation Form Cover Sheet - Patents Only. Please mail the recorded assignment document to the person whose signature, name and address appears at the bottom of this page.
11. The above checked items are being transmitted
 - a. before the 18th month publication.
 - b. after publication and the Article 20 communication but before 20 months from the priority date.
 - c. after 20 months but before 22 months (surcharge and/or processing fee included).
 - d. after 22 months (surcharge and/or processing fee included).

Note: Petition to revive (37 C.F.R. 1.137(a) or (b)) is necessary if 35 U.S.C. 371 requirements submitted after 22 months and no proper demand for International Preliminary Examination was made by 19 months from the earliest claimed priority date.

 - e. by 30 months and a proper demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date.
 - f. after 30 months but before 32 months and a proper demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date (surcharge and/or processing fee included).
 - g. after 32 months (surcharge and/or processing fee included).

Note: Petition to revive (37 C.F.R. 1.137(a) or (b)) is necessary if 35 U.S.C. 371 requirements submitted after 32 months and a proper demand for International Preliminary Examination was made by 19 months from the earliest claimed priority date.
12. At the time of transmittal, the time limit for amending claims under Article 19
 - a. has expired and no amendments were made.
 - b. has not yet expired.
13. Certain requirements under 35 U.S.C. 371 were previously submitted by the applicant on _____, namely:

SCHNADER HARRISON SEGAL & LEWIS

Date: 19 Jun 2001

By:

T. Ray Miller 31750

Austin R. Miller, Reg. No. 16,602
 1600 Market Street, 36th Floor
 Philadelphia, PA 19103

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Art Unit	:	<u>CUSTOMER NO. 22469</u>
Examiner	:	
Serial No.	:	
Filed	:	Herewith
PCT No.	:	PCT/JP00/07269
PCT Filed	:	October 19, 2000
Inventors	:	Koji Hanaoka Seiji Sakamoto Katsutoshi Igawa Yutaka Yamauchi Shizuki Kasaoka Toshiro Sawada Koichi Shinohara Yuji Tsukihara Shinjiro Baba
Title	:	BLAST FURNACE COKE HAVING HIGH REACTIVITY AND HIGH STRENGTH AND METHOD OF PRODUCING THE SAME

Docket No.: 1205-01

Confirmation No.:

Dated: June 19, 2001

PRELIMINARY AMENDMENT**Box PCT**

Assistant Commissioner for Patents
Washington, DC 20231

Sir:

Prior to action on the merits and in a format in accordance with the new rules, using a clean copy of the Specification and also a marked-up version of such, we respectfully request consideration of the following amendments and remarks:

In the Specification (Clean copy as amended)

Please replace the paragraph [0043] on page 12 with the following:

Further, as is seen in FIG. 5, it is apparent that a coal having medium rank and low fluidity (X) can obtain a target coke strength (TI_6 being approximately 84 %) which is an indicator as to whether it can be used in a blast furnace by being blended with 5 wt% to 40 wt% of each of caking coals (from A to F) thereto. When a caking coal is blended by less than 5 wt%, the strength becomes insufficient while, when the caking coal is blended

by more than 40 wt%, the strength exceeds the target value; however, since a larger quantity of high-priced caking coal is used, production cost becomes higher. Furthermore, it has become clear the higher the mean reflectance of caking coal becomes, the higher the improvement effect of coke strength becomes whereupon a larger quantity of coal having medium rank and low fluidity can be used.

Please replace Table 5 on page 23 with the following:

Table 5

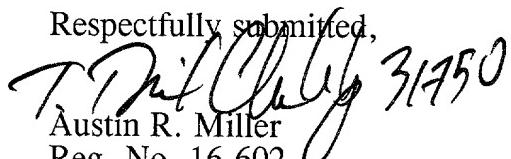
	Ratios of pores in respective pore size distributions (vol %)				$I_{RI=25\%}$	Strength TI_6
	less than 10 μm	less than 1 μm	10 μm to 100 μm	more than 100 μm		
Example 1	13	6	12	20	66.3	84.4
Example 2	13	6	11	24	65.4	84.3
Example 3	12	5	11	19	65.9	84.5
Example 4	12	4	12	24	65.1	84.3
Example 5	15	8	15	20	67.0	84.3
Example 6	12	6	10	15	68.1	84.6
Example 7	13	7	15	15	68.4	84.9
Comparative Example 1	12	6	16	24	63.2	84.1
Comparative Example 2	9	4	15	24	62.4	84.0
Comparative Example 3: Process coke	10	4	17	20	60.9	84.4

Remarks

We have amended the Specification to place it into better form for examination on the merits and allowance.

Passage to the appropriate art unit for examination on the merits is respectfully requested.

Respectfully submitted,



Austin R. Miller
Reg. No. 16,602
Attorney for Applicants

ARM:lh
(215) 563-1810

Version with Markings to Show Changes Made to the Specification

Please replace the paragraph [0043] on page 12 with the following:

Further, as is seen in FIG. 5, it is apparent that a coal having medium rank and low fluidity (X) can obtain a target coke strength (TI_6 being approximately 84 %) which is an indicator as to whether it can be used in a blast furnace by being blended with 5 wt% to 40 wt% of each of caking coals (from A to F) thereto. When a caking coal is blended by less than 5 wt%, the strength becomes insufficient while, when the caking coal is blended by [40 wt% or more] more than 40 wt%, the strength exceeds the target value; however, since a larger quantity of high-priced caking coal is used, production cost becomes higher. Furthermore, it has become clear the higher the mean reflectance of caking coal becomes, the higher the improvement effect of coke strength becomes whereupon a larger quantity of coal having medium rank and low fluidity can be used.

Please replace Table 5 on page 23 with the following:

Table 5

	Ratios of pores in respective pore size distributions (vol%)				$I_{RI=25\%}$	Strength Ti ₆
	less than 10 μm	less than 1 μm	10 μm to 100 μm	[100 μm or more] <u>more than 100 μm</u>		
Example 1	13	6	12	20	66.3	84.4
Example 2	13	6	11	24	65.4	84.3
Example 3	12	5	11	19	65.9	84.5
Example 4	12	4	12	24	65.1	84.3
Example 5	15	8	15	20	67.0	84.3
Example 6	12	6	10	15	68.1	84.6
Example 7	13	7	15	15	68.4	84.9
Comparative Example 1	12	6	16	24	63.2	84.1
Comparative Example 2	9	4	15	24	62.4	84.0
Comparative Example 3: Process coke	10	4	17	20	60.9	84.4

BLAST FURNACE COKE HAVING HIGH REACTIVITY AND HIGH STRENGTH AND METHOD OF PRODUCING THE SAME**Technical Field**

[0001] The present invention relates to blast furnace coke having high reactivity and high strength and a method of producing such blast furnace coke, and more particularly, to coke in which coke strength, reactivity with CO₂ and a pore size distribution are each at a desired level and a method of producing such coke.

Background of the Invention

[0002] In recent years, in view of an aging problem of a coke oven, there is an attempt to prolong a service life of the coke oven by decreasing an operating rate thereof. For such attempt, an operation of a blast furnace has been conducted while an injection quantity of pulverized coals was increased.

[0003] However, an increase of the injection quantity of pulverized coals into the blast furnace bring about not only a decrease of coke ratio but also an increase of load to coke in the blast furnace derived from an increase of an ore to coke ratio whereupon such increase of load exceeds coke strength to promote degradation of coke. Such degradation of coke aggravates gas permeability in the blast furnace to effect operational abnormalities such as hanging, slip and the like of a burden which, in the end, remarkably deteriorate an operation of the blast furnace. For this reason, it is important to suppress degradation of coke as much as possible.

[0004] Meanwhile, the coke in the blast furnace reacts with carbon dioxide (CO₂) to be partially gasified whereupon the coke becomes porous effecting a decrease of strength

thereof. As a method of suppressing such decrease of the strength thereof, a technique to reduce the reactivity with CO₂ has heretofore been under review; however, such technique causes an increase of energy cost of the blast furnace. Therefore, it is not advantageous from a standpoint of cost reduction and nowadays an operation with a low fuel ratio is rather required. In order to conduct such operation with a low fuel ratio, it is effective to pursue an increase of a reduction efficiency in the blast furnace by decreasing an operational temperature down to a temperature in a thermal reserve zone in a neighborhood of wustite-iron reduction equilibrium. For the reason described above, it is considered to use high reactive coke (CAMP-ISIJ, Vol. 5 (1992) 156).

[0005] Further, as a method of producing such high reactive coke, a method of increasing a ratio of non- to slight-caking coal in a material coal blend, a method of adding an inert coal material, that is, blending an inert substance as disclosed in Japanese Patent Laid-Open No. 313171/1994 and a method of blending char derived from a low carbonization coal as disclosed in Japanese Patent Laid-Open No. 117991 have been attempted.

[0006] However, as the blast furnace coke to be used under a circumstance in which the operation with a low fuel ratio as described above is required, a coke having characteristics that is high reactivity in a temperature region between a thermal reserve zone and a melting zone and also is hard to be degradation even after a reaction and another characteristic that is resistant to degradation in a temperature region between the melting zone to lower side of the furnace including a raceway section is required.

[0007] In relation with degradation characteristic of coke in the blast furnace, an understanding as described below has been prevailing. That is, as an index of this degradation characteristic, reactivity with CO₂ (CRI) and strength after reaction with CO₂

(CSR) are used and, particularly, CSR is regarded as important. Therefore, in blast furnace operations of Japanese iron and steel manufacturers, a management value is set on CSR and productions of coke have been conducted while maintaining a consistent CSR. However, as indicated by A line shown in FIG. 1, CRI and CSR are correlated with each other in a favorable manner; therefore, there has existed a problem that, if CSR is attempted to be maintained above a specified value, then CRI must be suppressed below another specified value. Plots in a neighborhood of the A line denote respective results of measurements of strengths after reactions on data prepared by varying a reaction time of process coke with CO₂ (at the time of CRI being 25 %, CSR being 60.9%).

[0008] In this regard, ordinary methods such as a method of increase a blending ratio of the non- to slight-caking coal, a method of adding an inert coal material and the like surely increase reactivity of coke, but, on the other hand, decrease melting capability between coal particles to effect decrease of coke strength; hence, it can not be said that they are effective methods of solving the above-described problems.

Disclosure of the Invention

[0009] An object of the present invention is to provide a blast furnace coke in which CO₂ reactivity is high and coke strength is large.

[0010] Another object of the present invention is to produce blast furnace coke having high reactivity and high strength at a low cost by using a coal blend composed of a small number of brands comprising a large quantity of semi-strong caking coal having medium rank and low fluidity (hereinafter referred to simply as “medium rank low fluidity coal”).

[0011] That is, the present invention proposes a blast furnace coke having high reactivity and high strength, the blast furnace coke being a coke that can be obtained by a method comprising the steps of:

charging a coal blend comprising 60 wt% or more of semi- heavy caking coal having medium rank and low fluidity in which a content of a non-melting inert component is 30 vol% or more in total into a coke oven; and

coking the coal blend,

characterized by having a pore size distribution in which a content ratio of pores having a diameter of less than $10 \mu\text{m}$ is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of from $10 \mu\text{m}$ to $100 \mu\text{m}$ is from 10 vol% to 15 vol%.

[0012] Preferably, the above-described pore size distribution is controlled such that a content ratio of pores having a diameter of less than $1 \mu\text{m}$ is 6 vol% or more and a content ratio of pores having a diameter of $100 \mu\text{m}$ or more is less than 20 vol%.

[0013] Further, the present invention is a blast furnace coke having high reactivity and high strength, the blast furnace coke being coke that can be obtained by the method comprising the steps of:

charging a coal blend comprising from 60 wt% to 95 wt% of semi-heavy caking coal having medium rank and low fluidity in which a mean reflectance (Ro) is from 0.9 to 1.1 and a maximum fluidity (MF) is 3.0 or less and the balance being a caking coal in which a mean reflectance (Ro) exceeds 1.1 into a coke oven; and

coking the coal blend,

characterized by having a pore size distribution in which a content ratio of pores having a diameter of less than $10 \mu\text{m}$ is from 12 vol% to 15 vol% and a content ratio of

pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol%.

[0014] Preferably, in the present invention, a caking coal in which a mean reflectance (Ro) is 1.3 or more and/or a semi-heavy caking coal in which a maximum fluidity (MF) is 3.0 or more is used as the balance of the above-described coal blend.

[0015] Preferably, further, the tumbler strength (wt% of +6 mm after 400 rotations; hereinafter referred to as "TI₆") is 83% or more.

[0016] Next, the present invention proposes a production method of a blast furnace coke having high reactivity and high strength characterized by comprising the steps of:

charging a coal blend comprising 60 wt% or more of semi-heavy caking coal having medium rank and low fluidity in which a content of a non-melting inert component is 30 vol% or more in total into a coke oven; and

coking the coal blend,

wherein the blast furnace coke is a coke having a pore size distribution in which a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of 10 μm to 100 μm is 10 vol% to 15 vol%.

[0017] Preferably, further in the present invention proposes a production method of a blast furnace coke having high reactivity and high strength characterized by comprising the steps of:

charging a coal blend comprising from 60 wt% to 95 wt% of semi-heavy caking coal having medium rank and low fluidity in which a mean reflectance (Ro) is from 0.9 to 1.1 and a maximum fluidity (MF) is 3.0 or less and the balance being a caking coke in which a mean reflectance (Ro) exceeds 1.1 into a coke oven; and

coking the coal blend,

wherein the blast furnace coke is a coke having a pore size distribution in which a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol%.

[0018] Preferably, further, in the present invention, a caking coal in which a mean reflectance (Ro) is 1.3 or more and/or a semi-heavy caking coal in which a maximum fluidity (MF) is 3.0 or more is used as the balance of the above-described coal blend.

[0019] Preferably, further, in the present invention, the tumbler strength TI_6 is 83 % or more.

[0020] Preferably, furthermore, the pore size distribution is controlled such that a volumetric content ratio of pores having a diameter of less than 1 μm is 6 vol% or more and a volumetric content ratio of pores having a diameter of 100 μm or more is 20 vol% or less.

[0021] According to the present invention, different from a conventional blend composed of a large number of brands in which 10 or more brands are blended, a blend composed of a small number of brands (about 5 brands or less) in which a large quantity of coal low in cost and abundantly available is blended can be realized whereupon a coke having a higher reactivity with CO_2 than an ordinary one and having coke strength equal to or higher than an ordinary one can be produced in a consistent manner.

Brief Description of the Drawings

[0022] FIG. 1 is a graph showing a relation between reactivity with CO_2 (CRI) and strength after reaction with CO_2 (CSR) of a conventional process coke;

[0023]FIG. 2 is a graph showing a relation between maximum fluidity (MF) and mean reflectance (Ro) of each coal;

[0024]FIG. 3 is a photograph of magnification power of 50 each of a single-brand coke made of coal having medium rank and low fluidity and an ordinary coke;

[0025]FIG. 4 is a graph showing an effect to be given to variation (ΔTI_6) of tumbler strength by a blend ratio between coal having medium rank and low fluidity and ordinary coal;

[0026]FIG. 5 is a graph showing an effect to be given to variation (ΔTI_6) of tumbler strength by a blend ratio between a coal having medium rank and low fluidity and a caking coal, and mean reflectance (Ro) of the caking coal;

[0027]FIG. 6 is a graph showing an effect to be given to $I_{\text{RI}=25\%}$ by a relation between volume of pores having a diameter of less than $1 \mu\text{m}$ and volume of pores having a diameter of $100 \mu\text{m}$ or more; and

[0028]FIG. 7 is a graph showing a relation between reactivity with CO_2 (CRI) and strength after reaction with CO_2 (CSR) of the coke according to the present invention.

Best Mode for Carrying Out the Invention

[0029]Inventors have studied, particularly, relations among pore morphologies, reactions and degradation characteristics of coke. That is, when diffusion of CO_2 into an inside of coke is considered, being based on an understanding that, when many fine pores are present therein, diffusion resistance of CO_2 is large; further, when surface areas of pores which are concerned with a gasification reaction are large, the above-described reaction with CO_2 is likely to center around the surface (the topochemical effect), components

contained in coal, above all, an inert component which has characteristics of holding fine pores even after the coal is coked is particularly noted.

[0030] Therefore, after coke is produced from a coal blend primarily composed of a coal having a large quantity of inert component, pore forms, reactivity and degradation characteristics thereof were investigated.

[0031] As a result, following knowledge was obtained: (1) coke produced from coal primarily composed of coal having a large quantity of inert component is rich in fine pores having a diameter of less than 10 μm , particularly less than 1 μm and has a large specific surface area; (2) the coke is relatively scarce in coarse pores having a diameter of from 10 μm to 100 μm , particularly 100 μm or more which are considered to affect coke strength; (3) when a number of the fine pores described in the above (1) is large, the reaction with CO_2 centers in such fine pores, which prevents the fine pores from becoming coarse, effectively acts on coke strength after reaction and enhances degradation resistance; (4) though the above-described topochemical effect is generated, degradation resistance as described in (3) is generated; and other characteristics.

[0032] Based on such knowledge, inventors have tried to produce coke having high reactivity and high strength.

[0033] That is, inventors have continuously studied blending of material coals. As a result, it was found that a “congeniality” in combination of brands, that is, a synergistic effect, exists in a specified brand of coal depending on a combination between a specified brand of coal and other brands of coal such that characteristics of coke derived from a coal blend based on blending of the specified brand of coal and other brands of coal have substantially been improved compared with characteristics, that is, a weighted mean value of strength,

reactivity with CO₂ and the like, of coke comprising a single brand derived from a coal comprising a single brand. With reference to this point, inventors have previously developed a method of estimating coke strength in which an interaction among brands has been taken into consideration (Japanese Patent Laid-Open No. 255066/1997).

[0034] It was confirmed that pore morphologies of the coke have a strong effect on this interaction and, accordingly, a coke having high reactivity and high strength is produced by making effective use of this congeniality.

[0035] Further, an observation of pore morphologies (or pore size distribution) and study of blending material coals have continuously been conducted. As a result, it was found that coke having high reactivity with CO₂ and high strength can be obtained, if only both content ratios of pores having a diameter of less than 10 μm and pores having a diameter of from 10 μm to 100 μm are controlled and, further, both content ratios of pores having a diameter of less than 1 μm and pores having a diameter of 100 μm or more are controlled.

[0036] To take an example, it was found that a pore size distribution as a characteristic of coke having high reactivity and high strength in which a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol%, preferably in addition thereto further a content ratio of pores having a diameter of less than 1 μm is 6 vol% or more, or a content ratio of pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol%, preferably in addition thereto further a content ratio of pores having 100 μm or more is 20 vol% or less is effective. In other words, since a specific surface area of pores having a diameter of less than 1 μm occupies 95% or more of a total, the higher the content ratio thereof becomes, the higher the reactivity becomes. On the other hand, since

relatively coarse pores having a diameter of 10 μm or more contribute to decrease of strength, the lower the content ratio thereof becomes, the higher the strength becomes (also after reaction).

[0037] Embodiments according to the present invention will now be described below along with steps which led to development of the present invention.

[0038] In the present invention, a coal blend having a blending ratio as high as 60% to 95% of coal having medium rank and low fluidity in which a non-melting inert content is 30% or more is carbonized in a coke oven. This coal having medium rank and low fluidity is classified as a semi-heavy caking coal petrographically. Though most of semi-heavy caking coals having a relatively high fluidity have 3.0 or more of maximum fluidity (MF) which is an indicator of caking property (encircled portion in FIG. 2), such coal having medium rank and low fluidity, as shown as shaded portion in FIG. 2, has lower maximum fluidity than the maximum fluidity and, further, a coal structure thereof, as shown in x coal and y coal described in Table 1, contains a large quantity of semi-fusinite, fusinite and the like which are inert components. Caused by this coal structure containing a large quantity of inert components, coke derived from this coal having medium rank and low fluidity is characterized by a large quantity of fine pores therein, as shown in FIG. 3.

[0039] Quality of coal having medium rank and low fluidity which is characterized as above has mean reflectance of 0.9 to 1.1 and maximum fluidity of 3.0 or less; such quality is approximately same as that (mean reflectance being about 1.07, maximum fluidity being 2.45) of a coal blend composed of multiple brands for use in an ordinary production of cokes. Nevertheless, inventors' study has found that, when two types of coal which are of approximately same quality, namely, a coal having medium rank and low fluidity and

an ordinary coal blend are mixed, coke strength has decreased, as shown in FIG. 4, though qualities of both types of coal are approximately same with each other, and that even a target coke strength can not be maintained.

[0040] Under these circumstances, inventors have further continued studies having in mind that an interaction, that is, an “congeniality” among brands of coals may be related with the above-described characteristics.

[0041] Particularly, coking tests have been conducted on coal blend prepared by blending a coal having medium rank and low fluidity (X) and several representative types of caking coals (from A to F) shown in Table 2.

[0042] Test results are shown in FIG. 5, where effects of blending ratios between a coal having medium rank and low fluidity and a caking coal and mean reflectance of the caking coal to strength (tumbler strength) of coke derived from respective coal blends are shown. In this regard, coke strength here means the above-described tumbler strength TI_6 ; an axis of ordinate in FIG. 5 shows improvement effects of tumbler strength of coke obtained by blending a coal having medium rank and low fluidity and caking coals (from A to F) under a condition that the coke strength of coke obtained by coking a single-brand coal having medium rank and low fluidity is set as 0. The axis of ordinate shows strength difference between a single-brand coke derived from coal having medium rank and low fluidity and a coke derived from a coal blend prepared by blending a coal having medium rank and low fluidity and a caking coal wherein the value 1.0 thereon represents an example of a process control target value. Further, numerals in FIG. 5 show blending ratios between a coal having medium rank and low fluidity and caking coals (from A to F). An axis of abscissa shows mean reflectance (Ro) of a caking coal.

[0043] Further, as is seen in FIG. 5, it is apparent that a coal having medium rank and low fluidity (X) can obtain a target coke strength (TI_6 being approximately 84%) which is an indicator as to whether it can be used in a blast furnace by being blended with 5 wt% to 40 wt% of each of caking coals (from A to F) thereto. When a caking coal is blended by less than 5 wt%, the strength becomes insufficient while, when the caking coal is blended by 40 wt% or more, the strength exceeds the target value; however, since a larger quantity of high-priced caking coal is used, production cost becomes higher. Furthermore, it has become clear the higher the mean reflectance of caking coal becomes, the higher the improvement effect of coke strength becomes whereupon a larger quantity of coal having medium rank and low fluidity can be used.

[0044] On this occasion, it was also clear that when a plurality of types of caking coals are used, that is, not confined to one type, same effect to the coke strength was obtained. An operation of preparing a coal blend in an actual coke production comes to be more efficient when less number of types of caking coals are used; in this regard, a number of types of caking coals may be set taking into consideration preparation operation time or inventories thereof; however, on thought of an ordinary operation, it is appropriate that a number of types of caking coals is from 1 to 3.

[0045] Ordinarily, since caking coal is an expensive type of coal, it can be said that it is desirous to suppress a blending ratio of this caking coal from a standpoint of coke production cost. Therefore, in the present invention, it is desirous to use at least one type of caking coal having mean reflectance of 1.3 or more which is highly effective in improving coke strength. In other words, this is because that use of caking coal having

mean reflectance of 1.3 or more shows an improvement effect only by a blending ratio of about 5 wt% to about 20 wt%.

[0046]The above-described coal having medium rank and low fluidity is classified petrographically as semi-heavy caking coal having similar mean reflectance, since it has mean reflectance Ro of 0.9 to 1.1; however, coal having medium rank and low fluidity has a relatively large quantity of an inert component among semi-heavy caking coal or compared with heavy caking coal having higher mean reflectance so that it is characterized by low fluidity. Ordinarily, coals soften-melt at a temperature between from 350°C to 550°C where the above-described inert component lacks melting performance and, further, the inert component itself is of a porous structure having fine pores whereupon, even when it becomes a semi-coke at a temperature between 550°C and 650°C after undergone a softening-melting treatment or, further, when it becomes product coke after being carbonized up to 1000°C, it not only holds the porous structure having fine pores but also permits a melting component to form fine pores and hold them. That is, a large quantity of fine pores are formed in the coke obtained by carbonizing the coal having a large quantity of inert component.

[0047]Next, measurement results of pore size distributions of a single-brand coke which can be obtained by coking only a coal having medium rank and low fluidity, a coke blend which can be obtained by a coking coal blend comprising a coal having medium rank and low fluidity and a caking coal, and an ordinary coke which can be obtained by coking a coal blend according to a blend composed of multiple brands are shown in Table 3. As is seen in Table 3, fine pores of less than 10 μm occupies a large share in the a single-brand coke obtained from only a coal having medium rank and low fluidity; to contrast,

a number of fine pores in a coke blend derived from a blend of coal having medium rank and low fluidity and a caking coal is a little less than the above but is larger than that of the ordinary coke. Further, in this case, a volume percent of relatively coarse pores of from 10 μm to 100 μm is smaller than that of the ordinary coke.

[0048] Next, steps as to how coke having high reactivity and high strength are prepared is described.

[0049] With reference to coke having high reactivity and high strength, there are many types of evaluation methods; in the present invention, 200 g of coke having a particle size of 20 mm \pm 1 mm are reacted up to 25 wt% thereof at 1100°C under a CO₂ flow with a flow rate of 5 l/min and the resultant reaction product is evaluated in terms of I-typed drum strength (percent by weight of +10 mm after 600 rotations) I_{RI=25%} whereupon coke in a relation of I_{RI=25%} \geq 65 is designated as coke having high reactivity and high strength.

[0050] As already described above, according to the present invention, different from a conventional coal blend composed of multiple brands comprising more than 10 brands of coals, even when a coal blend composed of a small number of brands (about 5 brands or less) blended with a large quantity of low-priced coal which is abundantly available is used, coke having reactivity with CO₂ higher than conventional coke and coke strength equal to or higher than the conventional one can be produced in a consistent manner.

[0051] In other words, the present invention can achieve the following:

- (1) A coal blend comprising 60 wt% or more of coal having medium rank and low fluidity in which a content ratio of inert component is 30 wt% or more in total is carbonized as a coke oven charge coal; further preferably;
- (2) A coal blend using a caking coal having 1.3 or more of mean reflectance (Ro)

and/or a semi-heavy caking coal having 3.0 or more of maximum fluidity (MF) as the balance of the coal blend described in the above (1) is carbonized as a coke oven charge coal; or

(3) A coal blend comprising 60 wt% or more of coal having medium rank and low fluidity in which mean reflectance (Ro) is from 0.9 to 1.1 and maximum fluidity (MF) is 3.0 or less is carbonized as a coke oven charge coal; further preferably

(4) A coal blend using a caking coal having 1.3 or more of mean reflectance (Ro) and/or a semi-heavy caking coal having 3.0 or more of maximum fluidity (MF) as the balance of coal blend described in the above (3) is carbonized as a coke oven charge coal.

[0052] Now, embodiments will be described below.

(1) Quality evaluation was conducted on a coke obtained from a coal blend comprising coal shown in Table 2.

[0053] A coke oven charge coal blend was prepared by using X coal as the above-described coal having medium rank and low fluidity which is a main component thereof, A coal as an example of coal having high carbonization for a purpose of reinforcement of strength, C coal as an example of a semi-heavy caking coal which exhibits mean reflectance equal to or higher than that of a semi-heavy caking coke having medium rank and low fluidity or a heavy caking coal at a blending ratio as shown in an expression of X coal : A coal : C coal = 81 : 9: 10.

[0054] Strength after reaction $I_{RI=25\%}$ at the time of reaction rate of 25% and coke strength TI_6 of coke derived from the above-described coal blend, namely, based on coal comprising a large quantity of coal having medium rank and low fluidity (hereinafter referred to simply as "coke of medium rank coal") were compared with those of an

ordinary coke derived from the ordinary coal blend thereby showing results in Table 4.

It became apparent that, though the coke of medium rank coal has coke strength TI_6 similar to that of the ordinary coke, it has an enhanced $I_{RI=25\%}$ compared with the ordinary coke.

That is, it became clear that it is a coke having high reactivity and high strength.

[0055] It is preferable to use, for example, Black water (BWR) coal produced in Australia as a coal having medium rank and low fluidity when such coke having high reactivity and high strength is produced.

(2) Next, a pore structure of coke having high reactivity and high strength will be described.

[0056] Cokes were prepared such that fine pores (diameters thereof being less than $10 \mu m$ and less than $1 \mu m$, respectively) and coarse pores (diameters thereof being from $10 \mu m$ to $100 \mu m$ and $100 \mu m$ or more, respectively) were varied in vol% and then respective pore size distributions were measured. Further, reactivity CRI, strength after reaction CSR, I-type drum strength (weight% of +10 mm after 600 rotations) at various reaction rates were measured to calculate $I_{RI=25\%}$ by means of a linear approximation techniques. Furthermore, tumbler strength TI_6 thereof were measured. Results of these measurements are shown in Table 5.

[0057] As shown in Table 5, when volume of pores having a diameter of less than $10 \mu m$ was from 12 vol% to 15 vol% and volume of pores having a diameter of from $10 \mu m$ to $100 \mu m$ was from 10 vol% to 15 vol% (Examples 1 to 7), $I_{RI=25\%}$ value was 65.0 or more and cold strength value TI_6 was comparable to that of an ordinary coke (process coke). On the other hand, when volume of pores having a diameter of less than $10 \mu m$ was not from 12 vol% to 15 vol% and volume of pores having a diameter of from $10 \mu m$ to $100 \mu m$

μm was not from 10 vol% to 15 vol% (Comparative Examples 1 to 3), $I_{\text{RI}=25\%}$ did not attain 65.0 or more.

[0058] Further, it was found that, as shown in FIG. 6, even among Examples 1 to 7, when volume of finer pores having a diameter of less than 1 μm was 6 vol% or more and volume of coarse pores having a diameter of 100 μm or more was 20 vol% or less (Examples 5 to 7), $I_{\text{RI}=25\%}$ was 66.0 or more and both reactivity and strength were high thereby effecting a coke which is hard to be pulverized. Furthermore, when volume of pores having a diameter of from 10 μm to 100 μm exceeded 15 vol%, and further, volume of pores having a diameter of 100 μm or more exceeded 20 vol% (Comparative Examples 1 and 2), TI_6 became lower.

[0059] From the above finding, it was found that a coke having high reactivity and high strength of $I_{\text{RI}=25\%}$ can be defined by a content ratio of fine pores having a diameter of less than 10 μm and a content ratio of coarse pores having a diameter of from 10 μm to 100 μm . Further, coke having higher reactivity and high strength can be defined by restricting a volumetric content ratio of pores having a diameter of less than 1 μm with reference to fine pores and a volumetric content ratio of pores having a diameter of 100 μm or more with reference to coarse pores. Accordingly, it was found that a blast furnace coke can be produced in an assured manner by controlling coke strength after reaction with CO_2 by means of a volumetric content ratio of fine pores having a diameter of less than 10 μm , preferably less than 1 μm and a volumetric content ratio of coarse pores having a diameter of from 10 μm to 100 μm and, further, that of coarse pores having a diameter of 100 μm or more.

[0060] Further, it was found that, as characteristics of a coke having high reactivity and

high strength, there is a pore size distribution in which a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol%, preferably that of pores having a diameter of less than 1 μm is 6 vol% or more, a content ratio of pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol% and, further, in addition thereto, a content ratio of pores having a diameter of 100 μm or more is 20 vol% or less.

(3) Production results of a blast furnace coke having high reactivity and high strength produced by employing a coal having medium rank and low fluidity will now be explained.

[0061] As is apparent from Examples 8 to 15 shown in Table 6, when a blending ratio of a coal having medium rank and low fluidity in which a quantity of inert component is 30 vol% or more was 60 mol% or more, cold strength TI_6 was 83.4 or more and strength after reaction $I_{\text{RI}=25\%}$ at a constant reaction rate of 25% was 65.0 or more whereupon such coal turned into a coke having high reactivity and high strength. Further, as shown in Examples 16 to 21, a coal comprising 60 wt% to 95 wt% of coal having medium rank and low fluidity in which mean reflectance (Ro) is from 0.9 to 1.1 and maximum fluidity (MF) is 3.0 or less and the balance being coal having mean reflectance (Ro) exceeding 1.1 turned also into coke having high reactivity and high strength in which TI_6 was 83.7 or more and $I_{\text{RI}=25\%}$ was 65.0 or more.

[0062] On the other hand, even when a content ratio of inert component was 30 vol% or more, if the blending ratio thereof was less than 60 wt% (Comparative Example 5), though cold strength TI_6 was more than that of a process coke (Comparative Example 4), $I_{\text{RI}=25\%}$ was 65.0 or less. Further, when a content ratio of inert component was less than 30 vol% (Comparative Examples 6 and 13), mean reflectance (Ro) was less than 0.9 (Comparative Example 7) or maximum fluidity (MF) exceeded 3.0 (Comparative Example 8) in a coal,

$I_{RI=25\%}$ was not 65.0 or more. Further, even when a blending ratio of coal having medium rank and low fluidity in which Ro was from 0.9 to 1.1 and MF was 3.0 or less was less than 60 wt% (Comparative Examples 9 and 10), $I_{RI=25\%}$ was a little larger than that of the process coke, but was not 65.0 or more. Further, when a blending ratio of coal having medium rank and low fluidity in which Ro was from 0.9 to 1.1 and MF was 3.0 or more was from 60 to 95 wt% and the balance of the coal had Ro of 1.1 or less (Comparative Examples 11 and 12), $I_{RI=25\%}$ was 65.0 or less.

[0063] As is apparent from the above description, it was found that a coke having high reactivity and high strength can be obtained by blending 60 wt% or more of a coal having medium rank and low fluidity in which a content ratio of inert component is 30 wt% or more or mean reflectance (Ro) is from 0.9 to 1.1, and maximum fluidity is 3.0 or less and the balance being a caking coal in which mean reflectance (Ro) is 1.3 or more and/or a semi-caking coal in which maximum fluidity (MF) is 3.0 or more.

[0064] An improvement effect of strength after reaction with CO_2 of the coke according to the present invention was investigated by varying a reaction rate and the thus investigated result is now explained. As is shown in FIG. 7, against ordinary A line (process coke), B line according to the present invention is a result (CSR = 67% at the time of CRI = 25%) of investigation on respective strengths after reaction of samples having different reaction rates prepared by changing reaction times of the coke in Examples 5; it is known that it is positioned above the ordinary line thereby permitting it to be a coke having high reactivity and high strength.

Industrial Applicability

[0065] In a production of an ordinary blast furnace coke, a method of blending a

multiplicity of brands where a coal blend is prepared by blending 10 or more brands of coals has been executed. By adopting the present invention, a coal having medium rank and low fluidity which has not easily been utilized in the method of blending a multiplicity of brands can affluently be used. Particularly, by controlling pore morphologies of fine pores which are originated in an inert component by means of blending an appropriate caking coal, a coke which can hold high coke strength even when reactivity with CO₂ is enhanced can be produced. As a result,

- (1) reduction of production cost of a blast furnace coke;
- (2) reduction of fuel cost of a blast furnace by enhancing reactivity with CO₂ of coke;
- (3) reduction of emission of CO₂ by lowering a ratio of being fired; and the like are effects among others which can provide a great merit not only to iron manufacturing industry but also to environmental protection.

Table 1

	Coal Properties				Coal macerals analysis		
	ASH	VM	MF	Ro	vitrinite (Vt)	semi-fusinite (SF)	fusinite (F)
A coal	7.9	29.5	4.17	1.12	70.2	9.5	3.6
B coal	8.7	20.4	2.63	1.49	82.7	5.2	7.5
C coal	9.1	28.3	3.91	1.12	78.2	8.6	4.7
D coal	8.9	18.6	1.72	1.60	80.1	8.9	1.9
E coal	9.3	24.2	2.08	1.19	78.0	5.5	10.6
F coal	8.6	35.7	2.45	0.83	65.3	17.0	3.9
X coal	7.6	28.2	2.40	1.05	51.0	46.0	1.5
Y coal	7.3	29.1	2.78	1.04	56.0	33.6	5.2

Table 2

	Mean reflectance Ro	Maximum Fluidity MF	Tumbler strength*) $\Delta T I_6$ (%)
Coal having medium rank and low fluidity - X	1.05	2.40	-
A	1.59	1.63	1.1
B	1.57	1.42	0.9
C	1.46	2.37	0.7
D	1.38	1.22	0.5
E	1.23	1.60	0.3
F	1.14	4.08	0.2

*) $\Delta T I_6$: Improved quantity of tumbler strength of a single-brand coke derived from a single-brand X coal at a blending ration of X coal/i coal (i = A to F) being 95/5.

Table 3

		Content ratio of pores having a diameter of less than 10 μm (vol%)	Content ratio of pores having a diameter of 10 μm to 100 μm (vol%)
Present method	Single-brand coke derived from coal having medium rank and low fluidity	15	14
	Coke blend derived from coal having medium rank and low fluidity	13	11
Comparative Example	Comparative Example: Ordinary coke	10	17

Table 4

	$I_{RI=25\%}$	TI_6
Ordinary coke	62.4	84.4
Coke derived from coal having medium rank and low fluidity	66.3	84.5

Table 5

	Ratios of pores in respective pore size distributions (vol%)				$I_{RI=25\%}$	Strength TI_6
	less than 10 μm	less than 1 μm	10 μm to 100 μm	100 μm or more		
Example 1	13	6	12	20	66.3	84.4
Example 2	13	6	11	24	65.4	84.3
Example 3	12	5	11	19	65.9	84.5
Example 4	12	4	12	24	65.1	84.3
Example 5	15	8	15	20	67.0	84.3
Example 6	12	6	10	15	68.1	84.6
Example 7	13	7	15	15	68.4	84.9
Comparative Example 1	12	6	16	24	63.2	84.1
Comparative Example 2	9	4	15	24	62.4	84.0
Comparative Example 3: Process coke	10	4	17	20	60.9	84.4

Table 6

	Coal having medium rank and low fluidity				Coal of the balance			$I_{RI=25\%}$	Strength TI_6
	Ratio	Ro	MF	TI	Ratio	Ro	MF		
Example 8	60	1.15	2.65	30	40	1.22	2.91	65.0	84.7
Example 9	60	1.15	2.65	30	40	1.30	2.87	65.2	84.8
Example 10	60	1.15	2.65	30	40	1.11	3.07	65.1	84.6
Example 11	80	1.15	2.65	30	20	1.22	2.91	65.7	84.2
Example 12	80	1.15	2.65	30	20	1.30	2.87	66.1	84.5
Example 13	80	1.15	2.65	30	20	1.11	3.07	65.8	84.1
Example 14	100	1.15	2.65	30	-	-	-	66.8	83.5
Example 15	100	1.12	2.40	33	-	-	-	68.9	83.4
Example 16	60	1.04	2.78	28	40	1.30	2.87	68.1	84.8
Example 17	60	1.04	2.78	28	40	1.11	3.07	65.0	84.5
Example 18	80	1.04	2.78	28	20	1.30	2.87	67.7	84.4
Example 19	80	1.04	2.78	28	20	1.11	3.07	65.8	84.2
Example 20	95	1.04	2.78	28	5	1.30	2.87	67.3	83.9
Example 21	95	1.04	2.78	28	5	1.11	3.07	67.0	83.7
Comparative Example 4: Process coke	-	-	-	-	-	-	-	62.4	84.4
Comparative Example 5	55	1.15	2.65	30	45	1.30	2.87	64.8	84.9
Comparative Example 6	100	1.05	2.75	20	-	-	-	63.1	83.5
Comparative Example 7	100	0.85	2.20	25	-	-	-	59.8	81.9
Comparative Example 8	100	1.02	3.21	25	-	-	-	63.2	83.4

Comparative Example 9	55	1.05	2.75	20	45	1.30	2.87	62.9	84.9
Comparative Example 10	55	1.05	2.75	20	45	1.11	3.07	63.4	84.5
Comparative Example 11	60	1.05	2.75	20	40	1.08	2.71	63.8	84.2
Comparative Example 12	95	1.04	2.78	28	5	1.08	2.71	60.9	83.5
Comparative Example 13	100	1.04	2.78	28	-	-	-	60.5	83.3

WHAT IS CLAIMED IS:

1. A blast furnace coke having high reactivity and high strength, the blast furnace coke being a coke that can be obtained by a method comprising the steps of:

charging a coal blend comprising 60 wt% or more of semi- heavy caking coal having medium rank and low fluidity in which a content of a non-melting inert component is 30 vol% or more in total into a coke oven; and

coking the coal blend,

characterized by having a pore size distribution in which a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol%.

2. A blast furnace coke having high reactivity and high strength, the blast furnace coke being coke that can be obtained by the method comprising the steps of:

charging a coal blend comprising from 60 wt% to 95 wt% of semi-heavy caking coal having medium rank and low fluidity in which a mean reflectance (Ro) is from 0.9 to 1.1 and a maximum fluidity (MF) is 3.0 or less and the balance being a caking coal in which a mean reflectance (Ro) exceeds 1.1 into a coke oven; and

coking the coal blend,

characterized by having a pore size distribution in which a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol%.

3. The coke as set forth in Claim 1 or 2, wherein the balance of said coal blend is characterized by a caking coal in which a mean reflectance (Ro) is 1.3 or more and/or a semi-heavy caking coal in which a maximum fluidity (MF) is 3.0 or more.

4. The coke as set forth in any one of Claims 1 to 3, wherein tumbler strength TI_6 is characterized by being 83% or more.

5. The coke as set forth in any one of Claims 1 to 4, characterized in that the pore size distribution is controlled such that a content ratio of pores having a diameter of less than $1 \mu\text{m}$ is 6 vol% or more and a content ratio of pores having a diameter of $100 \mu\text{m}$ or more is 20 vol% or less.

6. A production method of a blast furnace coke having high reactivity and high strength characterized by comprising the steps of:

charging a coal blend comprising 60 wt% or more of semi-heavy caking coal having medium rank and low fluidity in which a content of a non-melting inert component is 30 vol% or more in total into a coke oven; and

coking the coal blend,

wherein the blast furnace coke is a coke having a pore size distribution in which a content ratio of pores having a diameter of less than $10 \mu\text{m}$ is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of $10 \mu\text{m}$ to $100 \mu\text{m}$ is 10 vol% to 15 vol%.

7. A production method of a blast furnace coke having high reactivity and high strength characterized by comprising the steps of:

charging a coal blend comprising from 60 wt% to 95 wt% of semi-heavy caking coal having medium rank and low fluidity in which a mean reflectance (Ro) is from 0.9 to 1.1 and a maximum fluidity (MF) is 3.0 or less and the balance being a caking coke in which a mean reflectance (Ro) exceeds 1.1 into a coke oven; and

coking the coal blend,

wherein the blast furnace coke is a coke having a pore size distribution in which a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol%.

8. The production method as set forth in Claim 6 or 7, characterized by using a caking coal in which a mean reflectance (Ro) is 1.3 or more and/or a semi-heavy caking coal in which a maximum fluidity (MF) is 3.0 or more as the balance of said coal blend.

9. The production method as set forth in any one of Claims 6 to 8, wherein the tumbler strength TI_6 is characterized by being 83 % or more.

10. The production method as set forth in any one of Claims 6 to 9, characterized in that the pore size distribution is controlled such that a volumetric content ratio of pores having a diameter of less than 1 μm is 6 vol% or more and a volumetric content ratio of pores having a diameter of 100 μm or more is 20 vol% or less.

Abstract

Related to a blast furnace coke and a production method thereof. A coke having high reactivity and high strength in which coke strength, reactivity with CO₂ and a pore size distribution are each at a desired level can be produced at a low cost by using a coal blend composed of a small number of brands comprising a large quantity of caking coal having medium rank and low fluidity. The blast furnace coke, being obtained by charging a coal blend comprising 60 wt% or more of semi- heavy caking coal having medium rank and low fluidity in which a content of inert component is 30 vol% or more in total or another coal blend comprising 60 wt% to 95 wt% of semi-heavy caking coal having medium rank and low fluidity in which mean reflectance (Ro) is 0.9 to 1.1 and maximum fluidity (MF) is less than 3.0 and the balance being a caking coal in which mean reflectance (Ro) exceeds 1.1 into a coke oven and coking the coal blend, which has a pore size distribution where a content ratio of pores having a diameter of less than 10 μm is from 12 vol% to 15 vol% and a content ratio of pores having a diameter of from 10 μm to 100 μm is from 10 vol% to 15 vol% and the production method thereof.

09/868480

1/5

FIG. 1

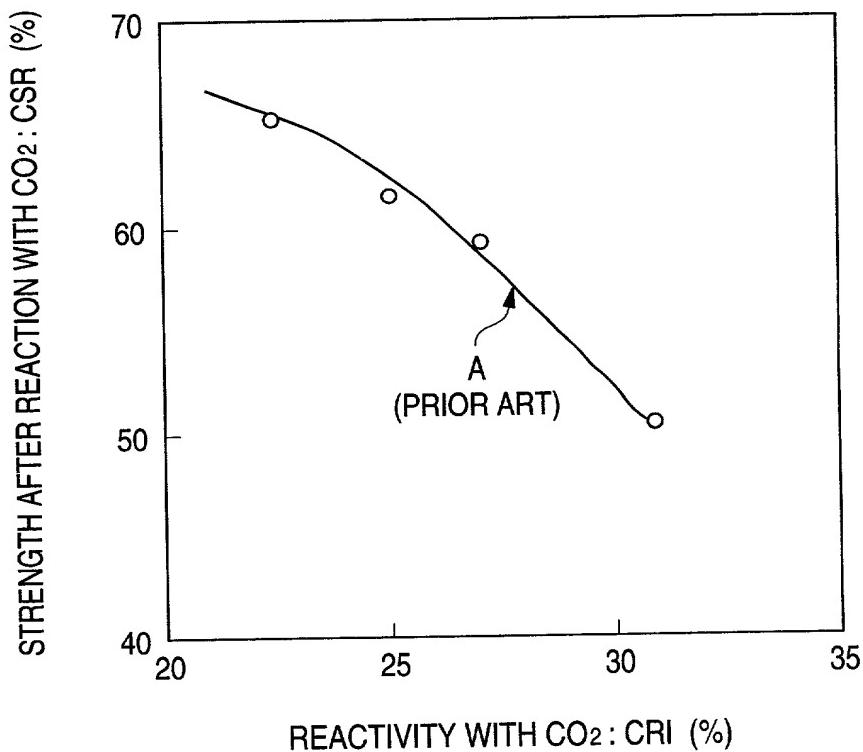
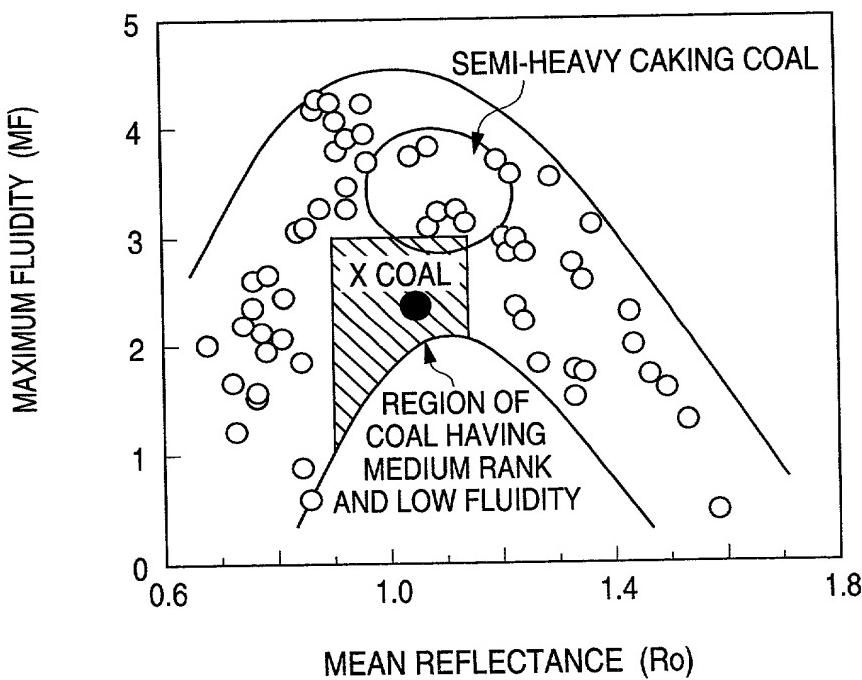


FIG. 2



2/5

FIG. 3A

SINGLE COKE OF COAL HAVING
MEDIUM RANK AND LOW FLUIDITY

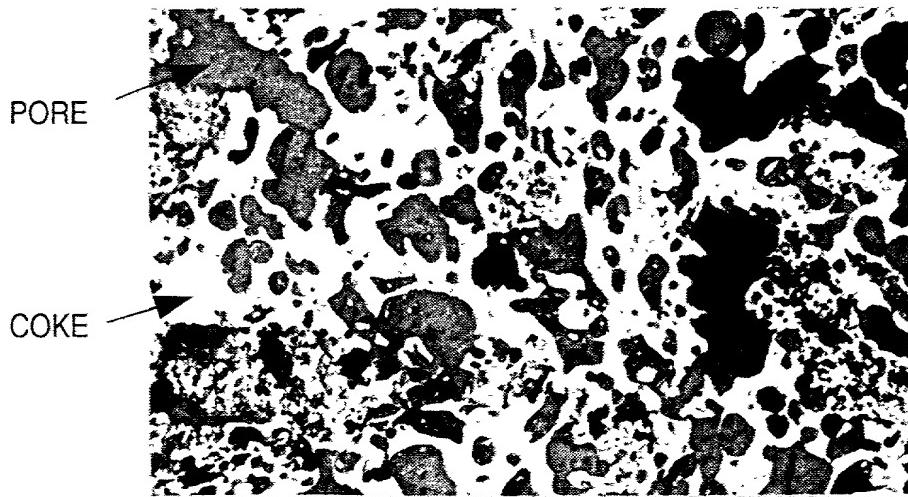
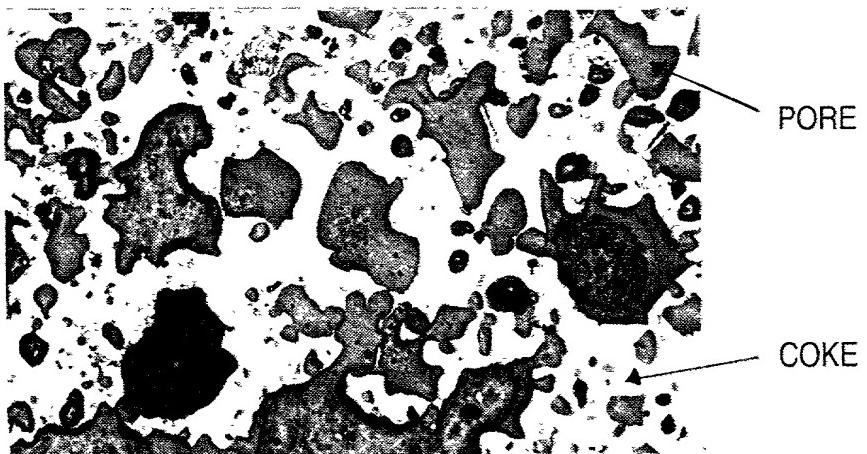


FIG. 3B

ORDINARY COKE



3/5

FIG. 4

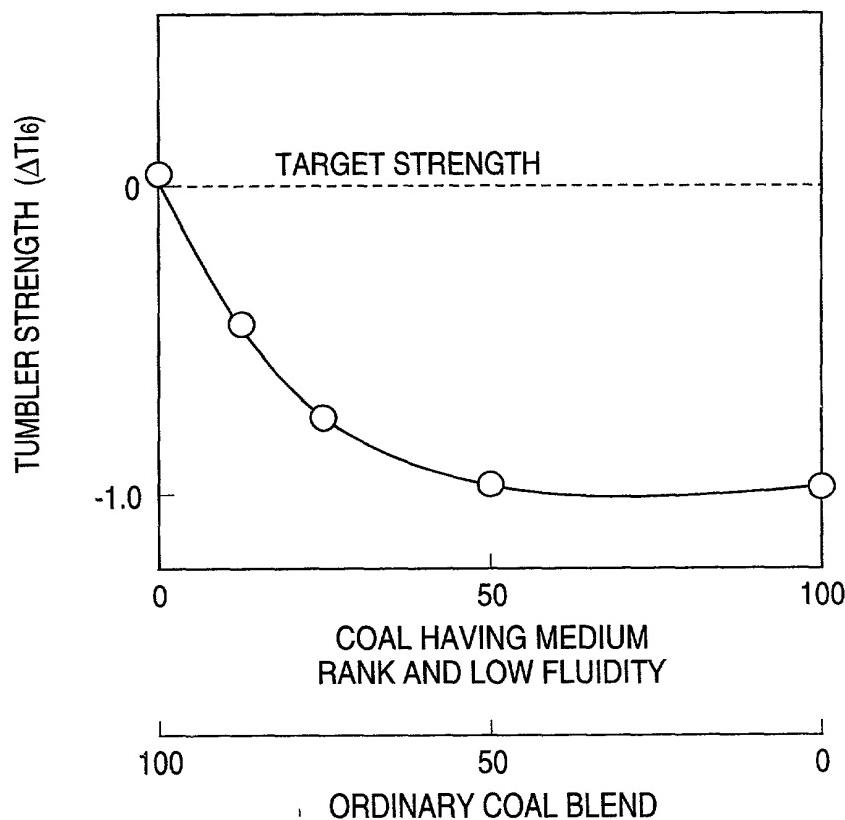
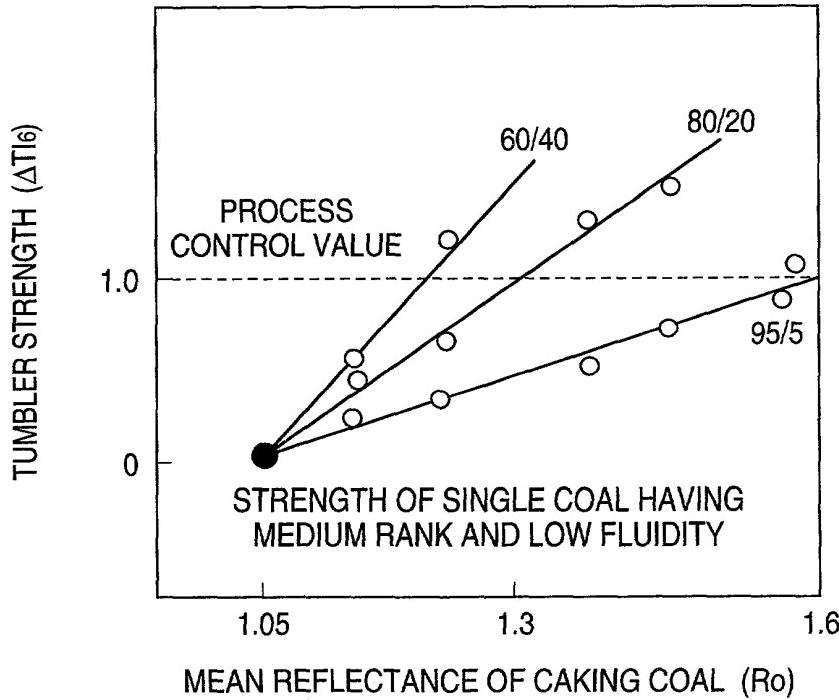
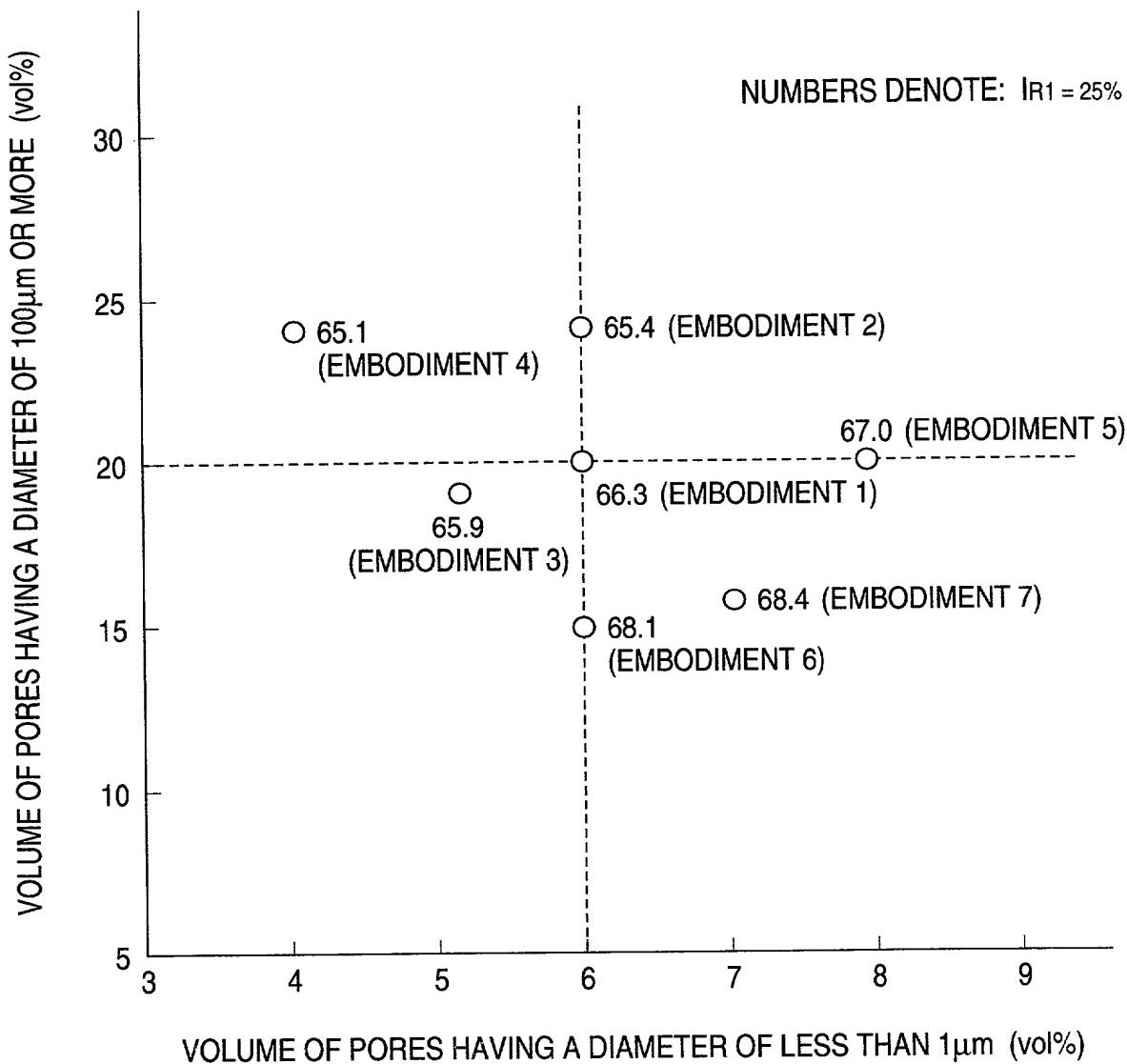


FIG. 5



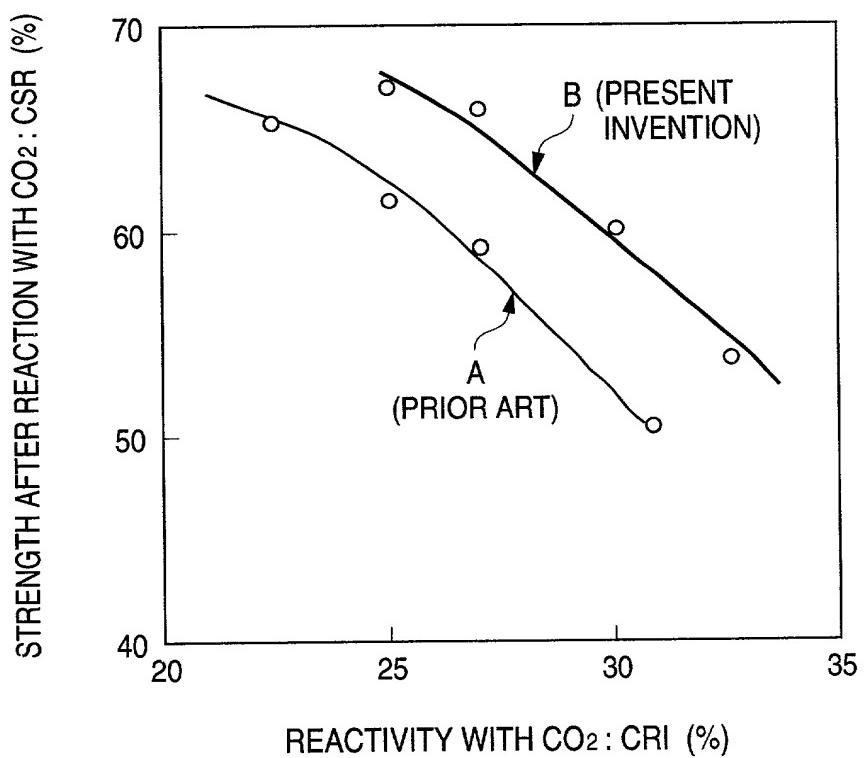
4/5

FIG. 6



5/5

FIG. 7



- Original Application
 PCT National Application
 U.S. Designated Office
 Continuation or Divisional Application
 Continuation-in-Part Application

**COMBINED DECLARATION,
POWER OF ATTORNEY AND PETITION**

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name,

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled BLAST FURNACE COKE HAVING HIGH REACTIVITY AND HIGH STRENGTH
AND METHOD OF PRODUCING THE SAME

which is described in the specification and claims

attached hereto.

filed on June 19, 2001

Application Serial No. 09/868,480

and was amended on _____

(if applicable)

which is described in International Application No. PCT/JPOO/07269

filed October 19, 2000 and as amended on

_____ (if any),

which I have reviewed and for which I solicit a United States patent.

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56.

I do not know and do not believe that this invention was ever known or used in the United States before my or our invention thereof or patented or described in any printed publication in any country before my or our invention thereof or more than one year prior to this application or said international application, or in public use or on sale in the United States of America more than one year prior to this application or said international application, or that the invention has been patented or made the subject of an inventor's certificate issued before the date of this application or said international application in any country foreign to the United States of America on an application filed by me or my legal representatives or assigns more than twelve months prior to this application or said international application, or that any application for patent or inventor's certificate on this invention has been filed in any country foreign to the United States of America prior to this application or said international application by me or my legal representatives or assigns except as identified below.

COMBINED DECLARATION, POWER OF ATTORNEY AND PETITION
(Page 2)

Attorney Docket No. 1205-01

I hereby claim foreign priority benefits under Title 35, United States Code, §119(a)-(d) or §365(b) of any foreign application(s) for patent or inventor's certificate, or §365(a) of any PCT International Application which designated at least one country other than the United States of America, listed below and have also identified below any foreign application(s) for patent or inventor's certificate or of any PCT International Application having a filing date before that of the application on which priority is claimed:

Number	Country	Date of Filing (day,month,year)	Priority Claimed
11-298609	Japan	20, 10, 1999	<input checked="" type="checkbox"/> yes <input type="checkbox"/> no
2000-278604	Japan	13, 09, 2000	<input checked="" type="checkbox"/> yes <input type="checkbox"/> no
			<input type="checkbox"/> yes <input type="checkbox"/> no
			<input type="checkbox"/> yes <input type="checkbox"/> no
			<input type="checkbox"/> yes <input type="checkbox"/> no

I hereby claim the benefit under Title 35, United States Code, §119(e) or §120 (as applicable) of any United States application(s) or §365(c) of any PCT International Application designating the United States of America, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT International Application(s) in the manner provided by the first paragraph of Title 35, United States Code, §112:

(Application Serial No.)

(Filing Date)

(Status) (patented, pending, abandoned)

(Application Serial No.)

(Filing Date)

(Status) (patented, pending, abandoned)

POWER OF ATTORNEY: As a named inventor, I hereby appoint the registered attorneys listed under Customer No. 022469 and the following registered attorneys to prosecute this application and transact all business in the United States Patent and Trademark Office connected therewith:

B
T. Daniel Christenbury Reg. No. 31,750
Guy T. Donatiello Reg. No. 33,167
Paul A. Taufer Reg. No. 35,703
James A. Drobile Reg. No. 19,690
Austin R. Miller Reg. No. 16,602
Gerard J. Weiser Reg. No. 19,763
Joan T. Kluger Reg. No. 38,940

Patrick J. Farley Reg. No. 42,524
Michael A. Patané Reg. No. 42,982
David A. Sasso Reg. No. 43,084
Robert A. McKinley Reg. No. 43,793
Sharon Fenick Reg. No. 45,269
Stewart M. Wiener Reg. No. 46,201

SEND CORRESPONDENCE TO:
IP Department
Schnader Harrison Segal & Lewis
36th Floor, 1600 Market Street
Philadelphia, PA 19103

DIRECT TELEPHONE CALLS TO
ATTORNEY OF RECORD AT:
(215) 563-1810

COMBINED DECLARATION, POWER OF ATTORNEY AND PETITION

(Page 3)

Attorney Docket No. 1205-01

I hereby petition for grant of a United States Letters Patent on this invention.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

1. FULL NAME OF SOLE OR FIRST INVENTOR <u>Koji Hanaoka</u>		INVENTOR'S SIGNATURE <i>Koji Hanaoka</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Technical Research Laboratories, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
2. FULL NAME OF JOINT INVENTOR, IF ANY <u>Seiji Sakamoto</u>		INVENTOR'S SIGNATURE <i>Seiji Sakamoto</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Technical Research Laboratories, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
3. FULL NAME OF ADDITIONAL JOINT INVENTOR, IF ANY <u>Katsutoshi Igawa</u>		INVENTOR'S SIGNATURE <i>Katsutoshi Igawa</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Technical Research Laboratories, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
4. FULL NAME OF ADDITIONAL JOINT INVENTOR, IF ANY <u>Yutaka Yamauchi</u>		INVENTOR'S SIGNATURE <i>Yutaka Yamauchi</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Mizushima Works, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
5. FULL NAME OF ADDITIONAL JOINT INVENTOR, IF ANY <u>Shizuki Kasaoka</u>		INVENTOR'S SIGNATURE <i>Shizuki Kasaoka</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Mizushima Works, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
6. FULL NAME OF ADDITIONAL JOINT INVENTOR, IF ANY <u>Toshiro Sawada</u>		INVENTOR'S SIGNATURE <i>Toshiro Sawada</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Mizushima Works, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
7. FULL NAME OF ADDITIONAL JOINT INVENTOR, IF ANY <u>Koichi Shinohara</u>		INVENTOR'S SIGNATURE <i>Koichi Shinohara</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Mizushima Works, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
8. FULL NAME OF ADDITIONAL JOINT INVENTOR, IF ANY <u>Yuji Tsukihara</u>		INVENTOR'S SIGNATURE <i>Yuji Tsukihara</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Mizushima Works, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			
9. FULL NAME OF ADDITIONAL JOINT INVENTOR, IF ANY <u>Shinjiro Baba</u>		INVENTOR'S SIGNATURE <i>Shinjiro Baba</i>	DATE June 11, 2001
RESIDENCE <u>Okayama, Japan</u>	<u>JPX</u>	CITIZENSHIP Japanese	
POST OFFICE ADDRESS <small>c/o Mizushima Works, Kawasaki Steel Corporation, Kawasaki-dori 1-chome, Mizushima, Kurashiki-shi, Okayama, 712-8074 Japan</small>			